

AD-A196 706

APPLICATIONS OF THE OPTICAL DIFFERENTIAL THERMAL  
ANALYSIS(U) ARMY LAB COMMAND MATERTOWN MA MATERIAL  
TECHNOLOGY LAB J L CASLAVSKY MAY 88 HTL-TR-88-18

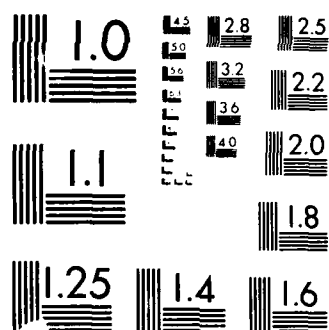
1/1

UNCLASSIFIED

F/G 20/13

NL





UTION TEST CHART  
 1.0 1.1 1.25 1.4 1.6 1.8 2.0 2.2 2.5 2.8 3.2 3.6 4.0 4.5 5.0 5.6 6.3

MTL TR 88-18

AD

2

DTIC FILE COPY

# APPLICATIONS OF THE OPTICAL DIFFERENTIAL THERMAL ANALYSIS

AD-A196 706

JAROSLAV L. CASLAVSKY  
CERAMICS RESEARCH BRANCH

May 1988

Approved for public release; distribution unlimited.



US ARMY  
LABORATORY COMMAND  
MATERIALS TECHNOLOGY LABORATORY

DTIC  
ELECTE  
JUN 28 1988  
S E D

U.S. ARMY MATERIALS TECHNOLOGY LABORATORY  
Watertown, Massachusetts 02172-0001

88 6 28 202

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

DD FORM 1473 EDITION OF 1 NOV 65 IS OBSOLETE

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

Block No. 20  
↓

## ABSTRACT

ODTA detects temperature contact-less, hence, it expands the upper usable temperature limit of DTA up to 3600°C accordingly suitable for studies of systems where enthalpic changes occur at high temperatures.

Application of the computer graphics for the evaluation of the ODTA curves is demonstrated on melting of  $Y_2O_3-Al_2O_3$  and  $YAl_3-GdAlO_3$  systems. Measurement errors occurring at temperatures above 1900°C are pointed out and possibilities for their elimination or at least diminution are discussed.

Accession For	
NTIS GRA&I	<input checked="" type="checkbox"/>
DTIC TAB	<input type="checkbox"/>
Unannounced	<input type="checkbox"/>
Justification	
By	
Distribution/	
Availability Codes	
Dist	Avail and/or Special
A-1	



UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

## INTRODUCTION

The Optical Differential Thermal Analysis (ODTA)<sup>1,2</sup> detects infrared radiation (IR) emitted by a sample heated inside the Black-Body Cavity (BBC). The BBC fulfills a dual function, it is the temperature source and the temperature reference standard.

BBC is a perfect radiator which emissivity is unity. The sample emissivity on the other hand, is a function of sample thermophysical and thermochemical properties, i.e., thermoradiative properties of the sample vary with the enthalpic changes taking place in the sample. In the great majority of cases changes of enthalpy and emissivity are in concert thereby enhancing the resolution of the inset of  $\Delta T_{max}$  and/or  $\Delta T_{min}$  peaks. This cooperation leads to a high sensitivity of the ODTA which allows  $\Delta T/t$  rates to be as low as 0.1°C/min. Consequently, phase relations can be studied at almost equilibrium conditions.

## ODTA APPARATUS

The ODTA apparatus consists of several parts: furnace with a power supply, furnace temperature control and a program system, two infrared pyrometers and a data acquisition system. With the exception of the furnace, most of the ODTA components are commercially available instruments. (The pyrometers are MAXLINE-SYSTEM M204, the furnace control and program system is MICRICON #82-300, manufactured by IRCON Inc. and RESEARCH Inc., respectively. The signal acquisition system consists of IBM-PC and IBM-AT.) For operation of the ODTA apparatus five independent microprocessors are required. Each has different digital characteristics, therefore, digital signals are converted to analog signals. To eliminate "cross talks", the D/A convertors are mutually optically insulated. Numerous sources of such instruments are available, hence, the selection of particular instrumentation depends on the operational range of the ODTA, and, on individual preference. The furnace is the critical part of the ODTA apparatus. The furnace design will be briefly discussed next.

The cross-section of the ODTA furnace, capable to operate in the 700 to 2200°C range in a vacuum or an inert gas, is shown in fig. 1.

A Black-Body Cavity *b* is designed as a hollow cylinder positioned by the heat sink *c*, concentrically with the heating element *e*. The heat sink is water-cooled as to provide sufficient heat removal from the BBC, especially at cooling rates higher than 1.0°C/min. The sample *a* is placed inside the BBC at such a height that it does not interfere with the optical path of the radiometer *d*. The maximum wavelength varies over the ODTA temperature range. The wavelength of the IR radiation is a nonlinear function of the temperature. Thus the IR radiation is converted to the temperature by a digital linearization using the Planck's law.

Only few materials do not undergo enthalpic changes in the 1500-3600°C temperature range. Among such materials are graphite and molybdenum, which both were found suitable for manufacturing of the BBC. A graphite BBC is shown in Fig. 1 right. It is a hollow cylinder, which walls are corrugated to enhance the absorption of IR radiation. In the center of the bottom is a sample holder, now replaced with the calibration reflector *f*. During measurements, the top of BBC is closed with a lid. The lid has a circular opening for measuring the sample temperature with the pyrometer *d<sub>s</sub>*. Pyrometer *d*, measures BBC temperature through an orifice in the side of the BBC (obscured in this view). Diameters of both orifices are calculated not to interfere with the optical path of either radiometer.

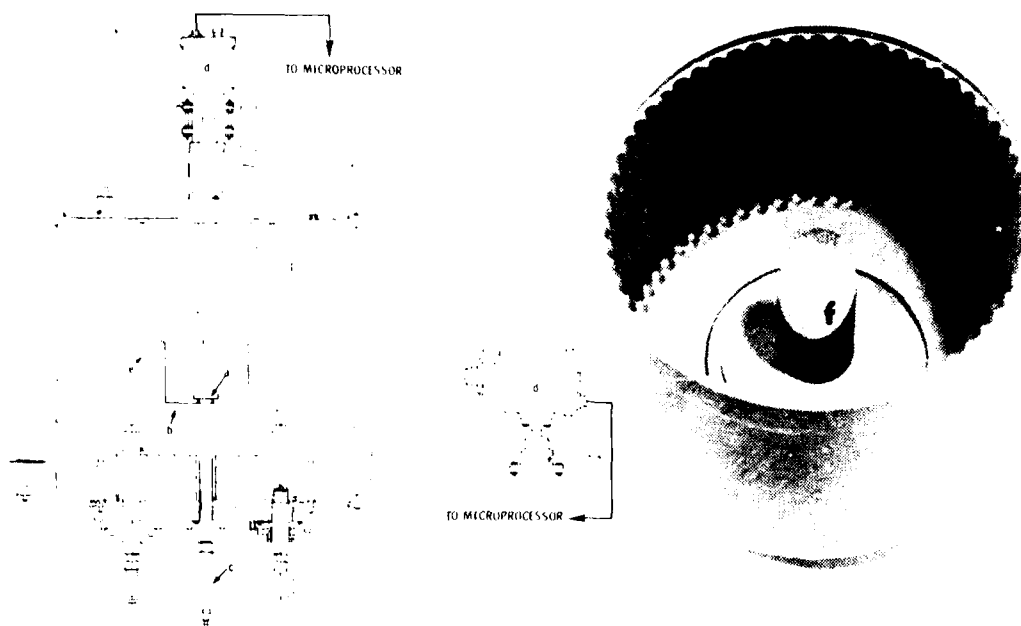


Fig. 1. Left: Cross-section of the ODTA furnace; right: The graphite Black-Body Cavity.

#### CALIBRATION

Two characteristics of optical pyrometers, the resolution power and the repeatability, will ultimately determine the accuracy of the ODTA. Optical pyrometers with the  $\pm 2^\circ\text{C}$  resolution power and the  $\pm 1^\circ\text{C}$  repeatability are commercially available. Yet, even such a good quality instrument will produce a low frequency noise in the ODTA. An attempt to filter such a noise by a common type of a filter will necessarily diminish the accuracy and the sensitivity of the ODTA measurements. The measurements are complicated by the impossibility to apply the Lambert's law accurately, and, to determine precisely the value of the shape factor  $A$ . Hence, these factors must be treated separately, and all such corrections included in a single correction factor.

Optical pyrometers are relative instruments and as such, they have to be calibrated. This is done by using materials with well known melting points. Two calibrated pyrometers are mounted on the ODTA furnace. Both are focused on the singular spot of reflector bloc  $f$  Fig. 1 right. The difference in temperature readings between pyrometers  $d_1$  and  $d_2$ , can be explained by the Lambert law. The difference is eliminated by setting the temperature reading of pyrometer  $d_2$  equal to that of pyrometer  $d_1$ , stepwise over the entire temperature range.

A new temperature difference will occur when the calibration reflector  $f$  is replaced by the sample holder. This time, the difference in the pyrometer readings is caused by a change of the shape factor  $A$ . The temperature difference caused by the shape factor and by other incalculable errors is called the integrated furnace background. It's value is determined experimentally and used for calculation of differential (DT) curves from the (T) curves.

## INTEGRATED FURNACE BACKGROUND

The heat transfer at temperatures over 1000°C occurs predominantly by radiation. If both, the heater and the sample are at thermal equilibrium, there is no heat transfer either way. During thermal analysis, the sample temperature has to be increased or decreased. In both the cases, a thermal gradient between the heater and the sample must be established. The degree of the gradient is determined by the  $\Delta T/t$  rate and the shape factor  $A$ . The degree of the gradient varies with temperature mainly due to the difference between the BBC and the sample emissivity. It is also effected by the furnace atmosphere. Values of these factors are particular to the furnace, the BBC and the sample materials i.e. to the setup of the entire ODTA apparatus. How these factors affect a (DT) curve is shown on the (DT) curve of melting of  $YAlO_3$  in Fig. 2.

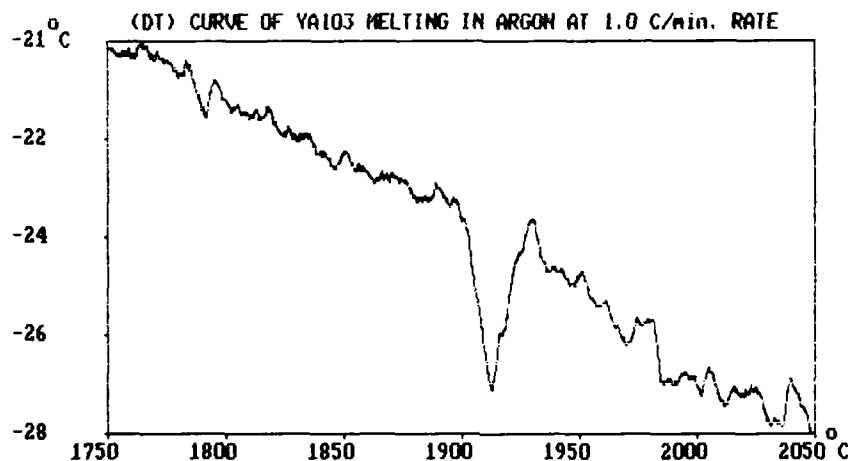


FIGURE 2.

For this reason, all the effectors are summarily determined and handled as an integrated furnace background (IFB). Due to the variability of these factors, the IFB has to be determined simultaneously with each individual measurement. For example, the (DT) curve shown in Fig. 2., now modified by the differentiation of the (T) curve with respect to the IFB, is shown in Fig.3.

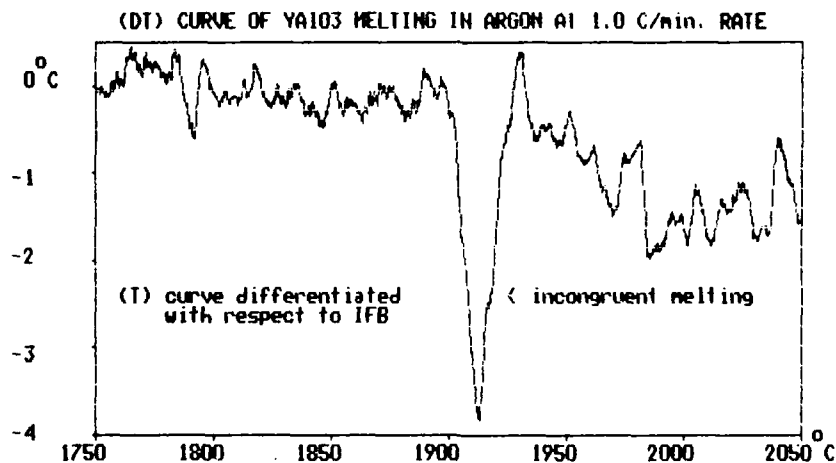
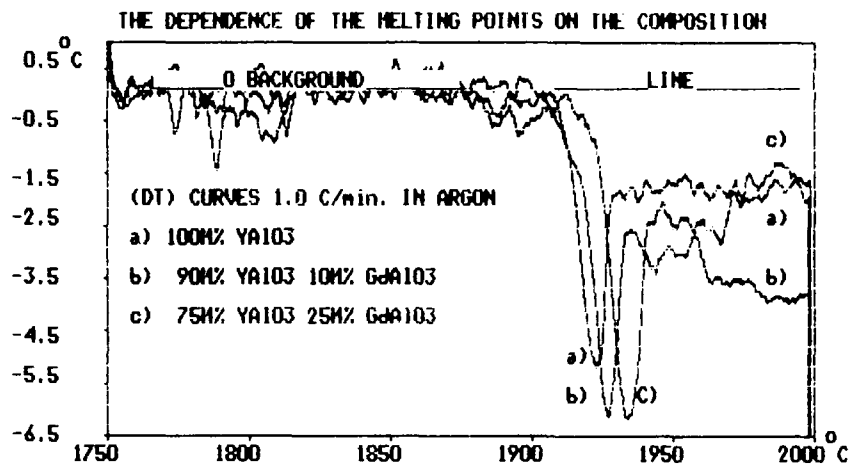


FIGURE 3.

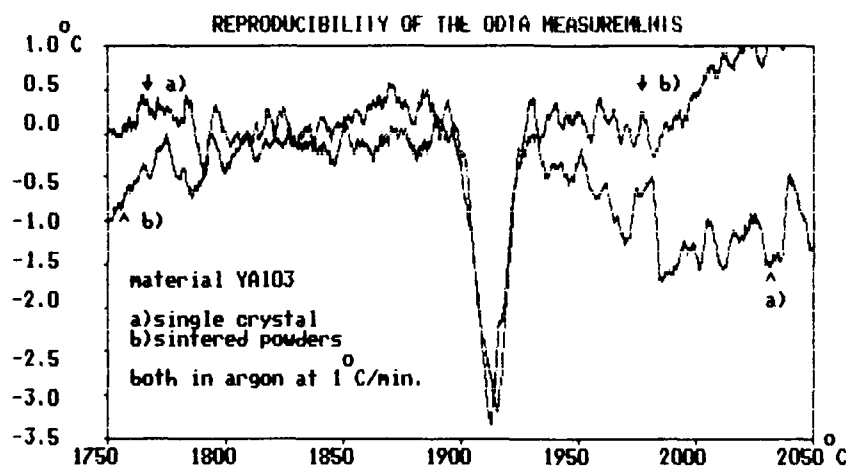


## SENSITIVITY AND REPRODUCIBILITY

Yttrium and gadolinium aluminates form solid solutions over a large compositional range. Melting points of individual single phases differ only slightly. This makes it difficult to determine a phase diagram by the conventional methods. Dependence of the melting points of individual members on composition, i.e., a)  $\text{YAlO}_3$ ; b) 90M%  $\text{YAlO}_3$  10M%  $\text{GdAlO}_3$  and c) 75M%  $\text{YAlO}_3$  25M%  $\text{GdAlO}_3$ , is shown in Fig. 4.



The lower end of the DTA measurement range was calibrated on the melting point of copper, while the upper end on the melting point of a single crystal of sapphire. The accuracy of the DTA measurement was found to be  $\pm 2^\circ\text{C}$  up to about  $1700^\circ\text{C}$ . Above this temperature, the error is larger. This is attributed to a particular melting behavior of sapphire<sup>1</sup>. The reproducibility of the measurements as determined from several hundred experiments is  $\pm 1^\circ\text{C}$ ; and is depicted in Fig. 5.



Curve a) in Fig. 5 is the same as in Fig. 3. Curve b) is from a sintered mixture of the powders. The apparent absence of an indication of an incongruent melting on curve b) is not attributed to the method, rather it shows that pre-melted samples give more detailed information.

### CONCLUSIONS

Previously, Rupert<sup>3</sup> used optical measurements to determine (DT) and (dT) curves, but his approach was different from the ODTA, as he used non-black body conditions and extremely high  $\Delta T/t$  rates. In any case, the choice in methods is commanded by the specificity of needs.

Materials important for ceramists and crystal growers are studied at temperatures not reachable by conventional DTA instruments. Therefore, at the present time, there is a lack of data for comparison with the data obtained by the ODTA. Distinct advantage of the ODTA is in the fact that it is not necessary to amplify the sensors' signals, i.e. the signals are not treated electronically. The only refinement of the ODTA data is mathematical. This approach allows to analyze the results using various aspects of mathematical evaluation and, in case of uncertainty, it is again simple mathematically to modify the process.

The ODTA zero line or background is effected by numerous factors. Nevertheless, certain characteristics of the zero line contain a significant diagnostic information; hence, before any evaluation of the data is done, a careful study of the background data has to be made. Because the ODTA is a new technique, used in only one laboratory so far, there are no text books on its use and just few publications with which to confront the results. However, we believe that the ODTA is quite a potent technique, opening new approach for studying materials at high temperatures.

### ACKNOWLEDGMENT

The author wishes to express his thanks to R. Allen and G. Bryant of MTL for their technical assistance and to E. Ramsden of the Boston University for the computer programming.

### REFERENCES

- 1 J. L. Caslavsky, Proceedings of the 9th International Congress for Thermal Analysis, Jerusalem, Israel, 1988.
- 2 J. L. Caslavsky, Report # MTL TR 88-11 U.S. Army Materials Technology Laboratory Watertown, Massachusetts 02172-0001
- 3 G. N. Rupert, Rev. Sci. Instr. 36, 1629 (1965).

# APPENDIX

An attempt is made here to refine the ODTA curves by a digital filter in order to view the  $\Delta T$  peaks in a grater detail. The digital filter does not cause any shift along the abscissa. This approach is aimed to the ODTA application to discern invariant and univariant points, i.e., to distinct a solid solution from an incongruently melting compound. At this time, however, there is an insufficient experimental support for a positive proof.

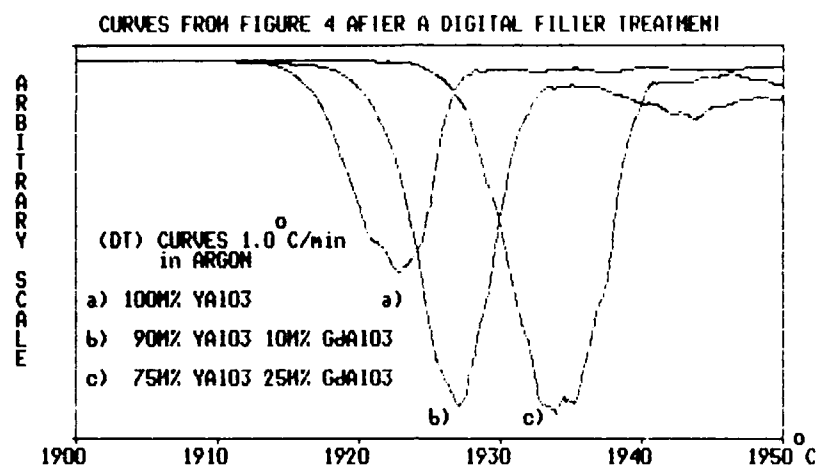


FIGURE 6.

## DISTRIBUTION LIST

No. of Copies	To	No. of Copies	To
	Office of the Under Secretary of Defense for Research and Engineering, The Pentagon, Washington, DC 20301		Director, U.S. Army Research & Technology Labs., Amer. Research Center, Mather Field, CA 94040
1	ATTN: Mr. J. Gersh	1	ATTN: DAVID-H, Dr. W. Carlson
1	Dr. L. Young	1	DAVID-AH, Dr. L. S. Stathis, AFRL-1,
1	Mr. E. R. Foster		Aeromechanics Laboratory
	Commander, U.S. Army Laboratory Command, 2800 Powder Mill Road, Adelphi, MD 20783-1146		Commander, U.S. Army Materiel Command, Redstone Arsenal, Huntsville, AL 35894-0001
1	ATTN: SECIS-IM-II	1	ATTN: AM-MIL-RD, Dr. J. B. Anderson
1	SECIS-ID	1	AM-MIL-EL, Dr. J. B. Anderson
1	SECIS-ID-A	1	AM-MIL-EL, Dr. J. B. Anderson
1	SECIS-PA	1	AM-MIL-EL, Dr. J. B. Anderson
1	SECIS-ID		Commander, U.S. Army Aviation Systems Command, Ft. Belvoir, IL 60005
	Commander, Defense Technical Information Agency Station, Building 5, 5410 Duke Street, Alexandria, VA 22304-6145	1	ATTN: AM-AV-CO, Mr. M. J. Baker
1	ATTN: DTIC-DW		Technical Library
1	National Technical Information Agency, 600 North Second Road, Springfield, VA 22161		Commander, U.S. Army Airborne Systems Command, Ft. Belvoir, IL 60005
	Director, Defense Advanced Research Projects Agency, 1400 Wilson Boulevard, Arlington, VA 22203	1	ATTN: AM-AV-CO, Mr. M. J. Baker
1	ATTN: Dr. P. Parrish		Technical Library
1	Dr. B. Wilcox		Commander, U.S. Army Satellite Communications Agency, Fort Monmouth, NJ 07030
1	Dr. K. Hardmann-Rhyne	1	ATTN: Technical Library
	Battelle Columbus Laboratories, Metals and Inorganic Information Center, 505 East Avenue, Columbus, OH 43260		Commander, U.S. Army Research and Technology Center, Fort Belvoir, IL 60005
1	ATTN: Mr. W. Duckworth	1	ATTN: Technical Library
1	Dr. D. Niesz		Commander, U.S. Army Communications and Electronics Command, Fort Monmouth, NJ 07030
	Department of the Army, Office of the Assistant Secretary of the Army (ROA), Washington, DC 20310	1	ATTN: AM-EL-TOR, Mr. T. A. Pfeiffer, Technical Lib.
1	ATTN: Dr. J. G. Prather, Dept. of Tech.		Director, Electronics Technology and Development Lab., Fort Monmouth, NJ 07030
1	Dr. J. R. Scully, SARO	1	ATTN: DTIC-DW, Dr. C. H. Denton
	Deputy Chief of Staff, Research, Development, and Acquisition, Headquarters, Department of the Army, Washington, DC 20310		Commander, U.S. Army Tank-Automotive Command, Warren, MI 48090
1	ATTN: DAMA-ZE, Mr. C. M. Church	1	ATTN: Dr. W. Bryzik
	Commander, U.S. Army Research and Development Office, Chief Research and Development, Washington, DC 20315	1	Dr. Rose
1	ATTN: Physical and Engineering Sciences Division	1	AM-TA-BE
	Commander, Army Research Office, P.O. Box 1111, Research Triangle Park, NC 27709-2211	1	AM-TA-LI, Technical Library
1	ATTN: Information Processing Office	1	AM-TA-R
1	Dr. G. Mayer		AM-TA-W, Dr. D. B. Dobbs
1	Dr. J. Hurt		Commander, U.S. Army Armament, Munitions, and Chemical Command, Dover, NJ 07801
1	Dr. A. Crowson	1	ATTN: Mr. J. Lannan
1	Dr. R. Reeber	1	Mr. R. F. Pebley, Jr., PLATH, Director
1	Dr. R. Shaw	1	Technical Library
1	Dr. R. E. Weigle	1	Dr. I. Davidson
1	D. Seitz	1	Dr. B. Ethana
	Commander, U.S. Army Materiel Command, 5401 Eisenhower Avenue, Alexandria, VA 22333	1	APMCM-HQ(D), Dr. J. T. Foxworth
1	ATTN: AMCLO, Dr. L. Hagan		Commander, U.S. Army Armament, Munitions, and Chemical Command, Rock Island, IL 61209
1	AMCDE, Mr. D. L. Griffin	1	ATTN: Technical Library
1	AMCQA-EQ, Mr. H. L. Light		Commander, U.S. Army Armament, Munitions, and Chemical Command, Aberdeen Proving Ground, MD 21010
1	AMCQA, Mr. S. J. Lorber	1	ATTN: AM-MU-PIN-I, Mr. J. Frank
	Commander, U.S. Army Electronics Research and Development Command, Fort Monmouth, NJ 07030		Commander, Aberdeen Proving Ground, MD 21010
1	ATTN: AMDET-ES, Dr. A. Tauber	1	ATTN: AMMCM-HQ(R), Mr. J. Vossler
	Director, Electronics Warfare Laboratory, Fort Monmouth, NJ 07030		U.S. Army Corps of Engineers, Construction Engineering Research Lab., P.O. Box 404, Champaign, IL 61810
1	ATTN: AMEW-G, Mr. M. Adler	1	ATTN: Dr. Robert Quattrone
	Commander, U.S. Army Materiel Systems Analysis Activity, Aberdeen Proving Ground, MD 21005		Commander, U.S. Army Belvoir ROME Center, Fort Belvoir, VA 20604-6006
1	ATTN: AMMY-MP, H. Cohen	1	ATTN: STIRL-IN, Mr. W. McGovern, Eval & Wt. App. Div.
	Commander, U.S. Army Night Vision Electro-Optics Laboratory, Fort Belvoir, VA 20600	1	AMOM-V, Mr. J. York
1	ATTN: DELAV-S, Mr. P. Travinsky	1	AMOM-HV, Dr. S. D. Steinbach
1	DELNV-L-D, Dr. R. Bauer	1	AMOM-JI, Mr. J. W. Loveless, Techn. Dir.
1	David P. Dr. L. Campbell	1	Mr. M. Taylor
	Commander, Barry Diamond Laboratories, 2800 Powder Mill Road, Adelphi, MD 20783		Director, U.S. Army Ballistic Research Laboratory, Aberdeen Proving Ground, MD 21010
1	ATTN: Technical Information Office	1	ATTN: AMMAR-Q, Dr. W. M. Smith
1	AMUL-PAE	1	AMMAR-HIT, Dr. W. M. Smith
		1	AMMAR-HIT, Dr. W. M. Smith
		1	AMMAR-HIT, Dr. W. M. Smith



No. of Copies	To
1	Georgia Institute of Technology, EIS, Atlanta, GA 30332
1	ATTN: Mr. J. D. Walton
1	Prof. R. Young
1	GTE Electronics, Waltham Research Center, 40 Sylvan Road, Waltham, MA 01981
1	ATTN: Dr. W. H. Rhodes
1	Martin Marietta Laboratories, 1450 South Rollins Road, Baltimore, MD 21242
1	ATTN: Dr. J. Venables
1	Massachusetts Institute of Technology, Department of Metallurgy and Materials Science, Cambridge, MA 02139
1	ATTN: Prof. J. H. Bowen
1	Prof. W. D. Kingery
1	Prof. J. Vander Sande
1	Materials Research Laboratories, P.O. Box 50, Ascot Vale, VIC 3032, Australia
1	ATTN: Dr. C. W. Weaver
1	Midwest Research Institute, 425 Wilkes Boulevard, Kansas City, MO 64110
1	ATTN: Mr. G. W. Gross, Head, Physics Station
1	Pennsylvania State University, Materials Research Laboratory, Materials Science Department, University Park, PA 16802
1	ATTN: Prof. P. Poy
1	Prof. R. E. Newman
1	Prof. R. E. Trexler
1	Dr. C. Pantano
1	Mr. C. O. Rood
1	Dr. R. E. Cross
1	Dr. M. Barnes
1	Dr. W. White
1	State University of New York at Albany, Department of Physics, Albany, NY 12222
1	ATTN: Prof. W. A. Linford
1	State University of New York at Stony Brook, Department of Materials Science, Long Island, NY 11790
1	ATTN: Prof. F. F. Y. Wang
1	Stanford Research International, 333 Ravenswood Avenue, Menlo Park, CA 94025
1	ATTN: Dr. P. Jorgensen
1	Dr. D. Rowcliffe
1	United Technologies Research Center, East Hartford, CT 06105
1	ATTN: Dr. J. Brennan
1	Dr. K. Frawe
1	University of California, Lawrence Livermore Laboratory, P.O. Box 808, Livermore, CA 94550
1	ATTN: Mr. R. Lindingham
1	Dr. C. E. Cline
1	Dr. J. Birch Holt
1	Dr. K. Newkirk
1	Dr. W. Knapke
1	University of Florida, Department of Materials Science and Engineering, Gainesville, FL 32611
1	ATTN: Dr. L. Hench
1	University of Washington, Ceramic Engineering Division, EE-10, Seattle, WA 98195
1	ATTN: Prof. R. Bragg
1	Westinghouse Electric Corporation, Research Laboratories, Pittsburgh, PA 15235
1	ATTN: Dr. R. J. Bratton
1	Battelle Pacific Northwest Lab, 901 Tenthon, Richland, WA 99363
1	ATTN: Mr. A. Burke, Associate Manager
1	Worcester Polytechnic Institute, Department of Materials Engineering, Troy, NY 12181
1	ATTN: R. J. Bratton
1	Oak Ridge National Laboratory, P.O. Box 116, Oak Ridge, TN 37831
1	ATTN: P. E. Beecher
1	V. J. Jennings
1	R. Johnson
1	Dr. J. Young

No. of Copies	To
1	Candor Laboratories, Allendale, NJ 07015
1	ATTN: Dr. E. Gentile, Tel 609-384
1	The Johns Hopkins University, Department of Civil Engineering, Materials Science and Engineering, Baltimore, MD 21202
1	ATTN: Dr. R. E. Green, Jr.
1	Director, Office of Science and Technology Policy, 945 Executive Office Building, Washington, DC 20503
1	Subcommittee on Science, 319 Rayburn House Office Building, Washington, DC 20515
1	ATTN: Mr. P. C. Maxwell
1	Aerospace Corporation, Materials Science Laboratory, 2550 East El Segundo Boulevard, El Segundo, CA 90245
1	ATTN: Dr. J. P. McCreight
1	IBM Corporation, Thomas B. Watson Research Center, Yorktown Heights, NY 10598
1	ATTN: Dr. G. Onoda
1	Corning Glass Works, Research and Development Division, Corning, NY 14830
1	ATTN: Dr. W. R. Brindley
1	3M Company, New Products Department, 219-01-01, 3M Center, St. Paul, MN 55144
1	ATTN: R. E. Richards
1	Technology Strategies, Inc., 10722 Shingle Oak Ct., Burke, VA 22015
1	ATTN: Dr. E. C. Van Reuth
1	Rutgers University, Center for Ceramics, Rm A274, P.O. Box 909, Piscataway, NJ 08854
1	ATTN: Prof. J. B. Wachtman, Jr., Director
1	Syracuse University, 304 Administration Building, Syracuse, NY 13210
1	ATTN: Dr. V. Weiss
1	Lehigh University, Materials Research Center #32, Bethlehem, PA 18015
1	ATTN: Dr. D. M. Smyth
1	Alfred University, New York State College of Ceramics, Alfred, NY 14002
1	ATTN: Dr. R. L. Snyder
1	University of California, Center for Advanced Materials, 050, Hilbrand Hall, Berkeley, CA 94720
1	ATTN: Prof. G. Somorjai
1	Boeing Aerospace Company, 11929 Southeast 901, Auburn, WA 98002
1	ATTN: W. E. Strobel
1	University of California, Materials Science and Mineral Engineering, Hearst Mining Building, Rm 204, Berkeley, CA 94720
1	ATTN: Prof. G. Thomas
1	University of Rhode Island, Kingston, RI 02881
1	College of Engineering
1	ATTN: Prof. Thomas J. Pickett
1	Hughes Research Laboratories, 9011 Malibu Road, Malibu, CA 90265
1	ATTN: Dr. A. J. Gentile
1	Dr. J. D. Parsons
1	Dr. A. Blazek, Department of Ceramic Technology, Institute of Technology, Wysoka Skola Technicka, Techniczna, Przemys, 6 Dobre, Bydgoszcz, 85-106
1	Prof. A. A. Chernov, Institute of Crystallography, Academy of Sciences of the USSR, Leninsky Prospekt 59, 117912 Moscow, USSR
1	Prof. Nicholas Ch. Egan, Dept. Ceramic Engineering, Center, 1900 NW Van Ness Avenue, Seattle, WA 98109
1	Dr. A. J. Sauerbrey, Center for Materials Research, Stanford University, Stanford, CA 94305

No. of Copies	To
1	Prof. M. Glicksman, Rensselaer Polytechnic Institute, Materials Engineering Department, Troy, NY 12181, USA
1	Prof. B. Henderson, Department of Physics, University of Strathclyde, John Anderson Building, 107 Rottenrow, Glasgow G4 0NG, Scotland UK
1	Dr. D. T. J. Hurle, RSRE, Malvern Words, WR 14 3 PS, UK
1	Prof. G. M. Imbusch, Department of Physics, University College, Galway, Ireland
1	Prof. E. Kaldis, Laboratorium fur Festkorperphysik, ETH-Honggerberg, CH-8093 Zurich, Switzerland
1	Prof. H. Komatsu, Research Institute of Iron, Steel and other Metals, Tohoku University, Sendai 980, Japan
1	Dr. R. A. Laudise, Bell Laboratories, 600 Mountain Avenue, Murray Hill, NJ 07974
1	Dr. K. Nassau, Bell Laboratories, 600 Mountain Avenue, Murray Hill, NJ 07974
1	Dr. Andrew D. Morrison, Jet Propulsion Laboratory, California Institute of Technology, 4800 Oak Grove Drive, Pasadena, CA 91109
1	Dr. Y. Shimony, Atomic Energy Commission, Nuclear Research Centre-NEGEV, POB 9001-Beer Sheva-Israel
1	Dr. Kenett Schwartz, Raychem Company, 300 Constitution Drive, Menlo Park, CA 94025
1	Dr. W. Tolksdorf, Philips Forschungslaboratorium, Vogt-Kolln-Str. 30, D-2 Hamburg 54, West Germany
1	Dr. S. Veprek, Anorgan Chem. Institut der Universitat, Winterthurerstr. 190, CH-8057 Zurich, Switzerland
1	Dr. B. Cocraine, Royal Signals and Radar Establishment, St. Andrews Road, Malvern, Words WR 14 3PS England
1	Dr. G. W. Cullen, RCA Laboratories, David Sarnoff Research Center, Princeton, NJ 08540
1	Prof. G. Huber, University of Hamburg, Applied Physics, Institute, 2000 Hamburg 36, West Germany
1	Prof. A. A. Kaminskii, A. V. Shubnikov Institute of Crystallography, Academy of Sciences of the USSR, SU-11 7333, Moscow USSR
1	Prof. P. F. Bongers, Nederlandse Philips Bedrijven B.V., Philips Research Laboratories, P.O. Box 60.000-5600 JA, Eindhoven, the Netherlands
1	Dr. G. Elliot, Hewlett Packard OED, 370 W. Trimble Rd., San Jose, CA 95131
1	Dr. P. A. Reynolds, Defense Advanced Research Projects Agency, 1400 Wilson Boulevard, Arlington, VA 22209
1	Dr. B. H. Kear, National Materials Advisory Board, Director of Center for Materials Synthesis, Rutgers University, P.O. Box 909, Piscataway, NJ 08854
1	Dr. H. L. Glass, Rockwell International, Electronic Devices Division, 3370 Miraloma Avenue, P.O. Box 3105, Anaheim, CA 92803

No. of Copies	To
1	Dr. L. DeShazer, Solid State Laser Associates, 16150 N.E. 85th Street, Suite 217 V, Redmond, WA 98052
1	Dr. P. Schlicht, Crystal Research, 1441 Sunnyside Terrace, San Pedro, CA 90732
1	Dr. T. Vastilo, University of Lowell, 1 University Avenue, Lowell, MA 01854-9985
1	Dr. N. Corbin, High Performance Ceramics, Goddard Road, Northborough, MA 01532-1545
1	Dr. L. J. Schioler, Aerojet Tech Systems Company, Dept. 9990, Bldg. 2019 - A2, P.O. Box 13222, Sacramento, CA, 95813
1	Prof. Menahem Strinberg, The Hebrew University of Jerusalem, Department of Inorganic & Analytical Chemistry, Jerusalem 91904, Israel
1	DuPont Company, 3411 Silver Road, Wilmington, DE 19899
1	ATTN: Library
1	Dr. F. S. Gill
1	Mellen Company, 10 Battle Road, Webster, NH 03303
1	ATTN: R. H. Mellen
1	Centorr Ass. Inc., Route 28, Suncook, NH 03275
1	ATTN: J. M. Lavoie
1	T. G. Jones
1	NASA Headquarters, Washington, DC 20456
1	ATTN: Dr. R. Crouch, Code EN
1	IBM T.H. Watson Research Center, York Town Height, P.O. Box 128, NY 10598
1	ATTN: Dr. E. Giess
1	The Gillitte Company, Gillette Park, Boston, MA 02106-2131
1	ATTN: Dr. C. H. Gajria
1	Rigaku USA, Inc., Northwoods Business Park, 200 Rose Wood Drive, Danvers, MA 01923
1	ATTN: Mr. Y. Sugiyama
1	Mr. M. Fornoff
1	Dr. A. G. C. Guzman, National Institute of Astrophysics, Optics, and Electronics, P.O. Box 216, 72000 Puebla, Pue., Mexico
1	Prof. A. Wold, Brown University, Department of Chemistry, Box H, Providence, RI 02192
1	IRCON, Inc., 7301 North Caldwell Avenue, Niles, IL 60648
1	ATTN: R. A. Goldberg
1	J. Salem
1	D. Williams
1	Philips Laboratory, 345 Scarborough Road, Briar Cliff, NY 13510
1	ATTN: W. Zwicker
1	G. Iacono
1	Director, U.S. Army Materials Technology Laboratory, Watertown, MA 02172-0001
1	ATTN: SMCMT-IM
1	Author

U.S. Army Materials Technology Laboratory,  
Watertown, Massachusetts 02172-0001  
APPLICATIONS OF THE OPTICAL DIFFERENTIAL  
THERMAL ANALYSIS - Jaroslav L. Caslavsky  
Technical Report MTL TR 88-18, May 1988, 9 pp -  
illus, D/A Project 1L161102AH42

AD  
UNCLASSIFIED  
UNLIMITED DISTRIBUTION

Key Words  
Thermal analysis  
Optical method  
Data acquisition

ODTA detects temperature contact-less, hence, it expands the upper usable temperature limit of DTA up to 3600°C accordingly suitable for studies of systems where enthalpic changes occur at high temperatures. Application of the computer graphics for the evaluation of the ODTA curves is demonstrated on melting of  $Y_2O_3-Al_2O_3$  and  $YAlO_3-GdAlO_3$  systems. Measurement errors occurring at temperatures above 1900°C are pointed out and possibilities for their elimination or at least diminution are discussed.

U.S. Army Materials Technology Laboratory,  
Watertown, Massachusetts 02172-0001  
APPLICATIONS OF THE OPTICAL DIFFERENTIAL  
THERMAL ANALYSIS - Jaroslav L. Caslavsky  
Technical Report MTL TR 88-18, May 1988, 9 pp -  
illus, D/A Project 1L161102AH42

AD  
UNCLASSIFIED  
UNLIMITED DISTRIBUTION

Key Words  
Thermal analysis  
Optical method  
Data acquisition

ODTA detects temperature contact-less, hence, it expands the upper usable temperature limit of DTA up to 3600°C accordingly suitable for studies of systems where enthalpic changes occur at high temperatures. Application of the computer graphics for the evaluation of the ODTA curves is demonstrated on melting of  $Y_2O_3-Al_2O_3$  and  $YAlO_3-GdAlO_3$  systems. Measurement errors occurring at temperatures above 1900°C are pointed out and possibilities for their elimination or at least diminution are discussed.

U.S. Army Materials Technology Laboratory,  
Watertown, Massachusetts 02172-0001  
APPLICATIONS OF THE OPTICAL DIFFERENTIAL  
THERMAL ANALYSIS - Jaroslav L. Caslavsky  
Technical Report MTL TR 88-18, May 1988, 9 pp -  
illus, D/A Project 1L161102AH42

AD  
UNCLASSIFIED  
UNLIMITED DISTRIBUTION

Key Words  
Thermal analysis  
Optical method  
Data acquisition

ODTA detects temperature contact-less, hence, it expands the upper usable temperature limit of DTA up to 3600°C accordingly suitable for studies of systems where enthalpic changes occur at high temperatures. Application of the computer graphics for the evaluation of the ODTA curves is demonstrated on melting of  $Y_2O_3-Al_2O_3$  and  $YAlO_3-GdAlO_3$  systems. Measurement errors occurring at temperatures above 1900°C are pointed out and possibilities for their elimination or at least diminution are discussed.

U.S. Army Materials Technology Laboratory,  
Watertown, Massachusetts 02172-0001  
APPLICATIONS OF THE OPTICAL DIFFERENTIAL  
THERMAL ANALYSIS - Jaroslav L. Caslavsky  
Technical Report MTL TR 88-18, May 1988, 9 pp -  
illus, D/A Project 1L161102AH42

AD  
UNCLASSIFIED  
UNLIMITED DISTRIBUTION

Key Words  
Thermal analysis  
Optical method  
Data acquisition

ODTA detects temperature contact-less, hence, it expands the upper usable temperature limit of DTA up to 3600°C accordingly suitable for studies of systems where enthalpic changes occur at high temperatures. Application of the computer graphics for the evaluation of the ODTA curves is demonstrated on melting of  $Y_2O_3-Al_2O_3$  and  $YAlO_3-GdAlO_3$  systems. Measurement errors occurring at temperatures above 1900°C are pointed out and possibilities for their elimination or at least diminution are discussed.



END

DATE

FILMED

DTIC

9-88